7th Grade Science Test (9/22) Review Questions

- 1. What's atomic number?
- 2. What's atomic mass number?
- 3. What is the name of the element with a symbol of Hg? Al? Si? Xe? H?
- 4. What's the name of the last family on the Periodic Table?
- 5. Which scientist came up our current model of the Periodic Table?
- 6. Which element has the lowest atomic number?
- 7. Given an atomic number of 12 and an atomic number of 30, how many protons, electrons, and neutrons are there in that element?
- 8. What is an isotope?
- 9. What is an alloy?
- 10. What properties do metals have?
- 11. What properties do non-metals have?
- 12. Why Alkali metals are very active?
- 13. What are valence electrons?
- 14. Why noble gases are inert?
- 15. What is a metalloid?
- 16. What's the name of the process of going from solid to gas? Liquid to solid? Liquid to gas?
- 17. What causes matter to change state?
- 18. How many quarks are needed to make a proton?
- 19. What do we call the gas form of water?
- 20. What's the density of an object if the volume is 1000 cm³ and the mass is 300 g?
- 21. What's the volume of an object if the density is 0.3 g/cm³ and the mass is 100 g?
- 22. What's the mass of an object if the density is 1.6 g/cm³ and the volume is 200 cm³?
- 23. What chemical properties do Alkali Earth metals have?
- 24. What chemical properties do halogens have?
- 25. What chemical properties do transitional metals have?
- 26. What is ionization energy?
- 27. Na has very high chance to react with halogens. Why?
- 28. John Dalton came up with a model for the atoms, what's the name of the model?
- 29. Who discovered electrons?
- 30. What's the significance of gold foil experiment done by Ernest Rutherford?